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$\text{NH}_4^+$  and HCN by comparing their ionization constants. The ionization constant of HCN is given in Table E1 as  $4.9 \times 10^{-10}$ . The ionization constant of  $\text{NH}_4^+$  is not listed, but the ionization constant of its conjugate base,  $\text{NH}_3$ , is listed as  $1.8 \times 10^{-5}$ . ...

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