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Separation of Multicomponent Mixture of Naphthalene ...

Separation of a mixture of naphthalene derivatives by reversed-phase high-performance liquid chromatography with UV detection was studied. Chromatographic characteristics were calculated for nine compounds, including naphthalene.

Separation of Multicomponent Mixture of Naphthalene ...

The invention relates to chemical technology of organic intermediates used in the production of various derivatives of naphthalene used as dyes, surfactants and other substances. The basis for many such derivative is 2-naphthalenesulfonate.

Method of naphthalene sulphonation

The Chemistry and Technology of Naphthalene Compounds. N. Donaldson; ... The Recent Advances in Application and Synthesis of Naphthalene Compounds. Article.

Naphthalene Derivatives - ResearchGate

Naphthalene | C10H8 | CID 931 - structure, chemical names, physical and chemical properties, classification, patents, literature, biological activities, safety ...

Naphthalene | C10H8 - PubChem

Naphthalene, the simplest of the fused or condensed ring hydrocarbon compounds composed of two benzene rings sharing two adjacent carbon atoms; chemical formula, C 1 0 H 8. It is an important hydrocarbon raw material that gives rise to a host of substitution products used in the manufacture of dyestuffs and synthetic resins.

Naphthalene | chemical compound | Britannica

Naphthalene isomers are comprised of two fused benzene rings. The naphthalene isomers are the dominant PAHs in fresh crude petroleum (Figures 5.3 A and G) and tar (Figure 5.3 G). The naphthalene isomer pattern of fresh crude oil serves as a reference sample for identifying individual isomers within each homolog pattern (Figure 5.6 A). Many isomers are identified based on available authentic compound standards and literature references.

Naphthalene - an overview | ScienceDirect Topics

As expected, compounds such as the benzene dicarboxylic acids having aromatic character and carbonyl groups show great similarity in their spectra) patterns to the naphthalene dicarboxylic acids. In the range 260-310 nm benzene-1,2-dicarboxylic acid has \nax 280 nm and benzene-1,3-dicarboxylic acid Am ax 280 nm with another peak at 288 nm (in aqueous solutions)9.

Characterization of naphthalene dicarboxylic acids. Part 1 ...

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Naphthalene - - Major Reference Works - Wiley Online Library

Naphthalene is the most abundant polycyclic aromatic hydrocarbon (PAH) found in urban air. It is reactive in the atmosphere under ambient conditions, its chief reaction partner being the hydroxyl radical, OH •.

Reaction of Naphthalene and Its Derivatives with Hydroxyl ...

Donaldson, N.: The Chemistry and Technology of Naphthalene Compounds; Edward Arnold, London (1958) Google Scholar

Naphthalin — Herstellung und Verwendung | SpringerLink

Hydrolysis is not expected to be an important environmental fate process since this compound lacks functional groups that hydrolyze under environmental conditions. Occupational exposure to 1,6-dimethylnaphthalene may occur through inhalation and dermal contact with this compound at workplaces where 1,6-dimethylnaphthalene is produced.

1,6-Dimethylnaphthalene | C12H12 - PubChem

Naphthalene is an organic compound with formula C 10 H 8. It is the simplest polycyclic aromatic hydrocarbon, and is a white crystalline solid with a characteristic odor that is detectable at concentrations as low as 0.08 ppm by mass.

Naphthalene - Wikipedia

Laser-induced fluorescence detection was used to measure photolysis rates of anthracene and naphthalene at the air–ice interface, and the kinetics were compared to those observed in water solution and at the air–water interface. Direct photolysis proceeds much more quickly at the air–ice interface than at the air–water interface, whereas indirect photolysis due to the presence of ...

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