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reaction with reactants on the left, products on the right, and an arrow separating the two. Small whole number ratios that are placed in front of the formulas in an equation in order to balance it.

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CHAPTER 11 CHEMICAL REACTIONS

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11.1—Describing Chemical Reactions

*How do you write a word equation? To write a word equation, write the names of the reactants to the left of the arrow separated by a plus sign; write the names of the products to the right of the

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arrow, also separated by plus signs.

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Reactions(continued) 9. Use the symbols in Table 11.1 on page 323 to write a skeleton equation for the following chemical reaction. Hydrochloric acid reacts with zinc to produce aqueous

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SECTION 11.1 DESCRIBING CHEMICAL REACTIONS (pages 321–329) This section explains how to write equations describing chemical reactions using appropriate symbols. It also describes how to write balanced chemical equations when given the names or formulas of the reactants and products in a chemical reaction.

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SECTION 11.1 DESCRIBING CHEMICAL REACTIONS (pages 321-329)

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Section 11.1 - Describing Chemical
Reactions In a chemical reaction, the
reactants are written on the left and the
products on the right. The arrow that
separates them is called yield. Reactants
□ Products

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As shown in Figure 11.3.1, applying a small amount of heat to a pile of orange

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Ammonium dichromate powder results in a vigorous reaction known as the ammonium dichromate volcano. Heat, light, and gas are produced as a large pile of fluffy green chromium(III) oxide forms. We can describe this reaction with a chemical equation. An expression that gives the identities and quantities of the ...

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Question 1. Give an example for complex chemical reaction. Answer: 'Photo synthesis' is a complex chemical reaction. Question 2. Give an example of simple for chemical reactions. Answer: Rusting of iron and combustion of fuels are simple chemical reactions. Question 3.

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equilibrium constant.

Chapter 11: Reactions and Other Chemical Processes ...

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Study Workbook CHAPTER 11, Chemical
Reactions(continued) 9 Use the symbols
in Table 111 on page 323 to write a

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Answers

skeleton equation for the following
chemical reaction Hydrochloric acid
reacts with zinc to produce aqueous
zinc(II) chloride and hydrogen

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